



# Transformational Technologies: Approach and Successes

**David Luebke**

Technical Coordinator for  
Carbon Capture

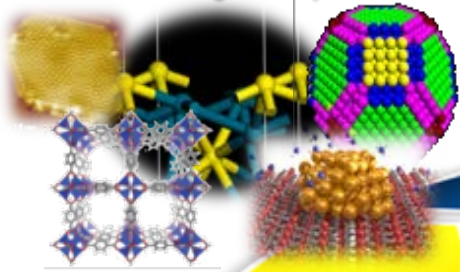
July 30, 2014



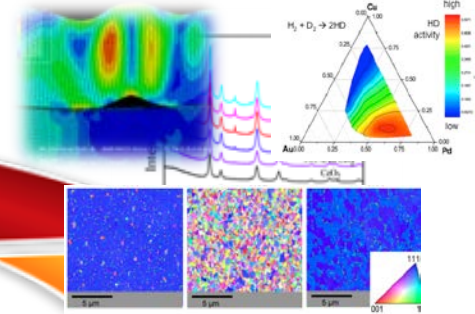
# Integrated Technology Development

*Material Synthesis & Fabrication*

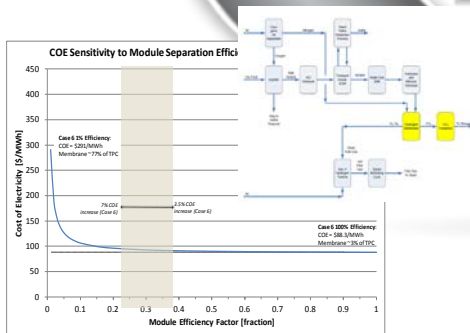
*Molecular Design & Optimization*



*Characterization*



*Accelerating Discovery,  
Development & Deployment*



*Process Synthesis &  
Techno-economic  
assessment*

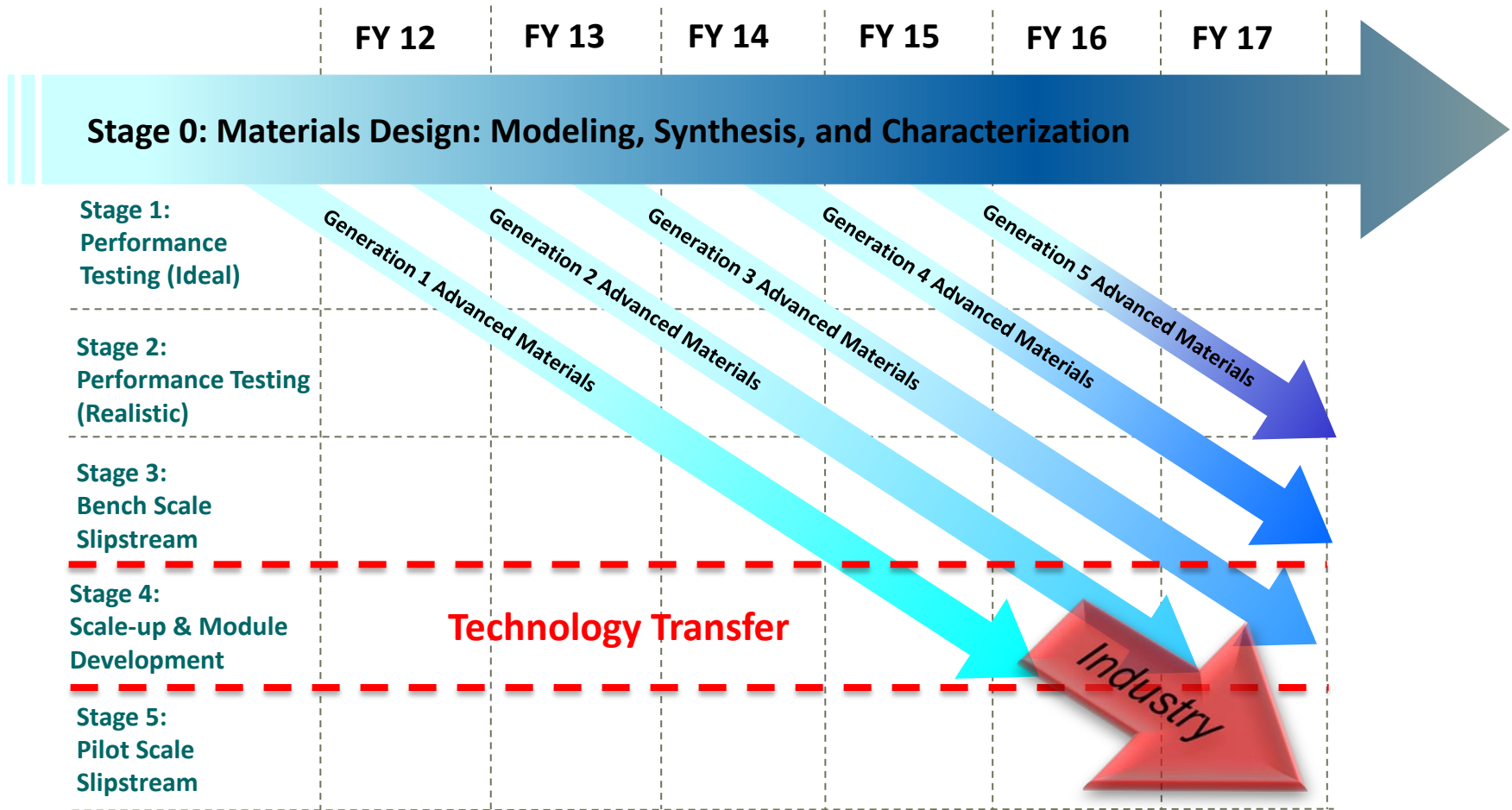


*Performance Assessment  
In Real Environments*



*Material Processing & Device  
Development*

# Integrated Technology Development Technology Pathway



# What Is the NETL-ORD Role in Transformation Technology Development?

## And what is it not?

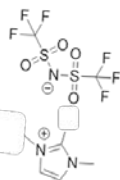
- **It's not:**
  - Basic science
  - Creation/discovery of new classes of materials
  - Pilot-scale testing
  - Commercialization
- **It is:**
  - Examination of novel classes of materials for capture
  - Exploration of innovative process configurations
  - Development of advanced screening approaches



# Materials

# Eutectic Solvents

- ILs forming crystalline solids tend to have a sharp melting point and low viscosity in the liquid phase.
- These materials also tend to melt well above room temperature.



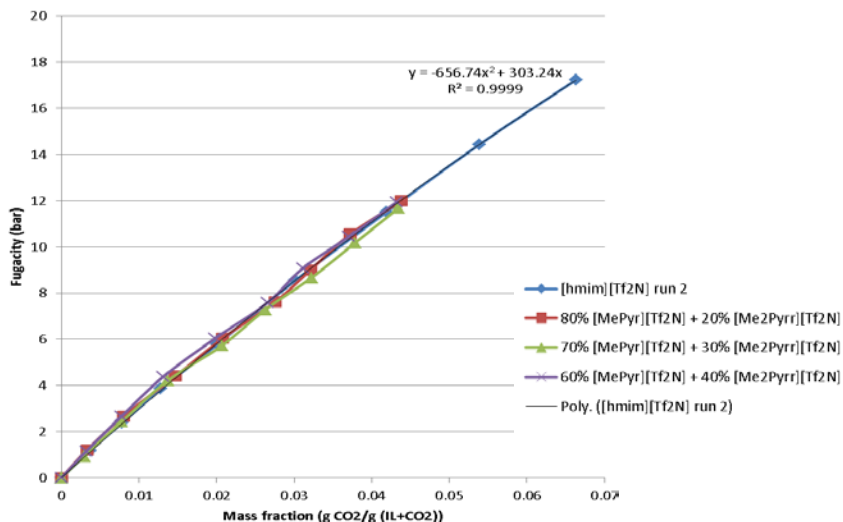
Identify Simpler compounds (Minimize interactions)

Eutectic

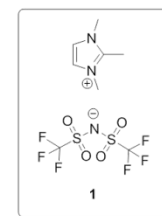
Suppression of  $T_m$

Low viscosity

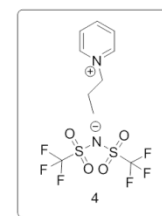
- Weakened ionic interaction
- Packing/Defect



- Forming eutectic mixtures could lead to low viscosity liquids.
- It proved challenging to locate mixtures showing both reduced viscosity and a depressed melting point.

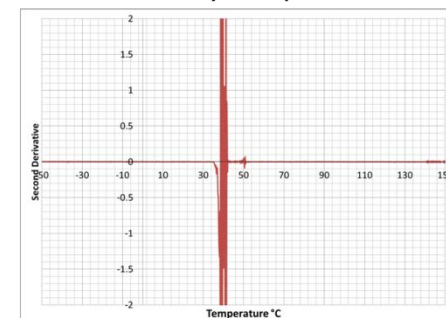


106°C



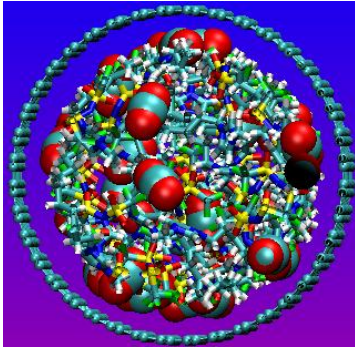
46.5°C

Viscosity is 22 Cp

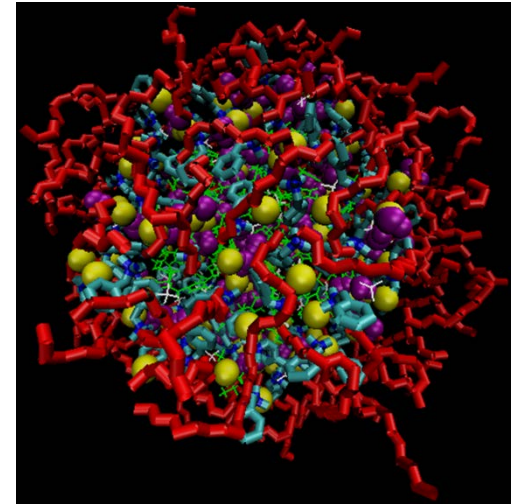


# Structured Liquids

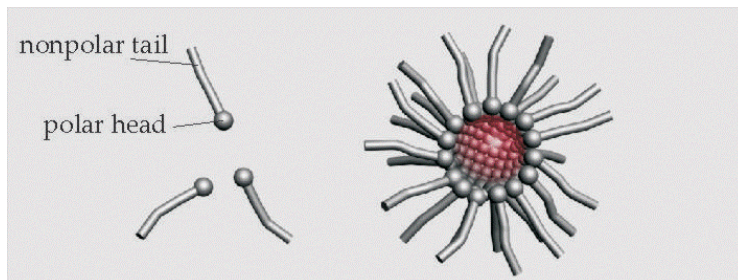
## IL Confinement



- ILs for semi-ordered structures when confined in pores under 50 nm.
- Structures show unique properties not attainable in bulk ILs.
- Formation of IL micelles results in similar property changes.
- Computational results appeared promising, but fabrication of the materials proved challenging.



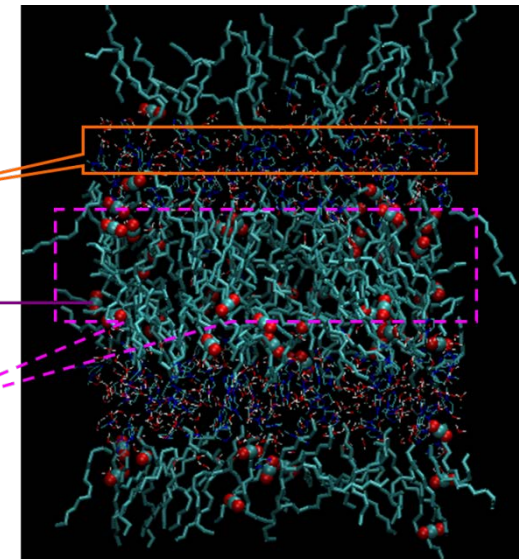
## Unconfined Structured ILs



Hydrophilic  
Head Groups

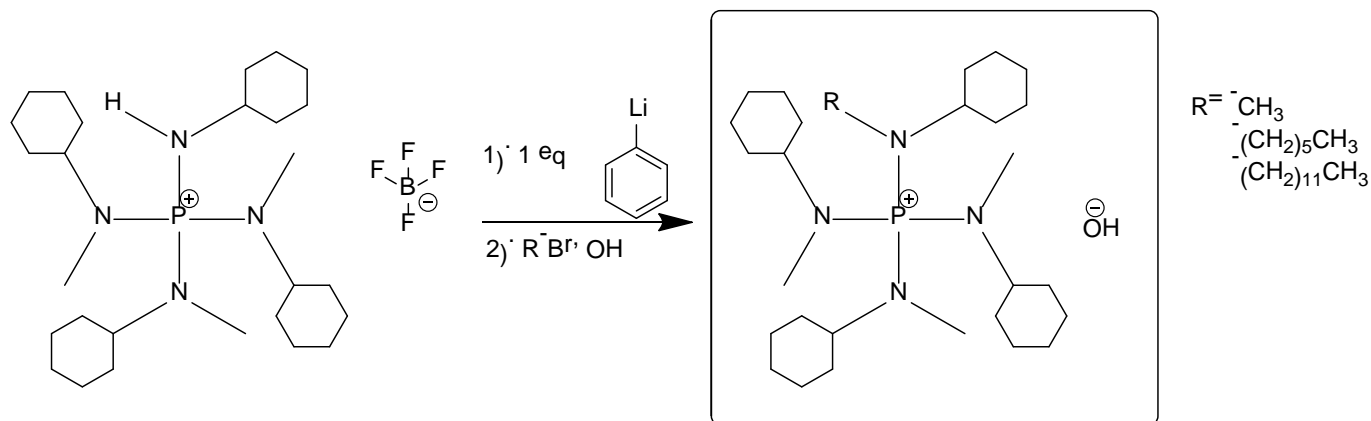
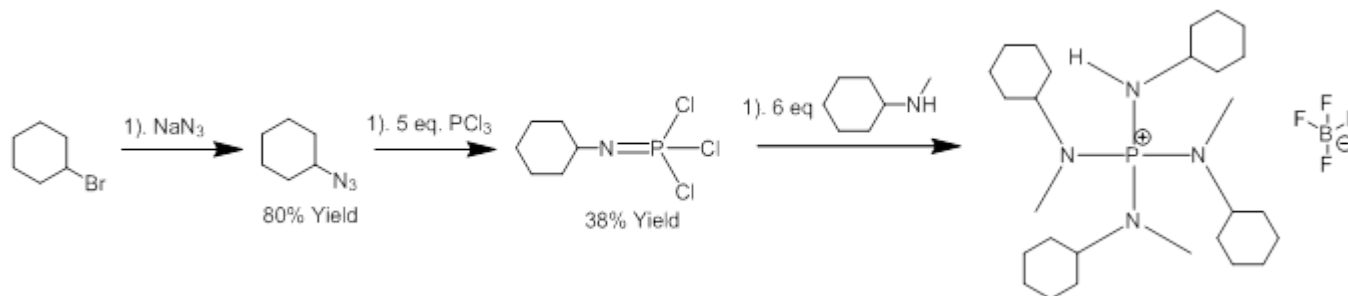
CO<sub>2</sub>

Hydrophobic  
Tails





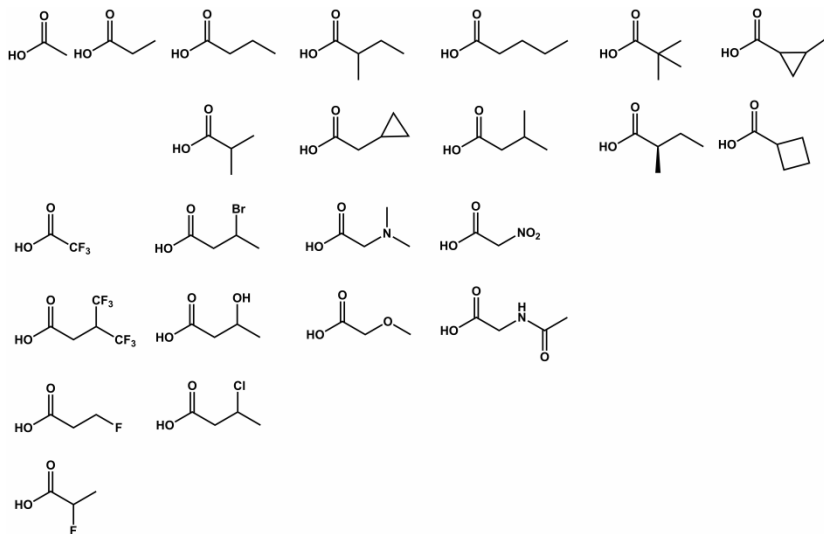
# Hybrid Organic-Inorganic Hydroxide Solvents



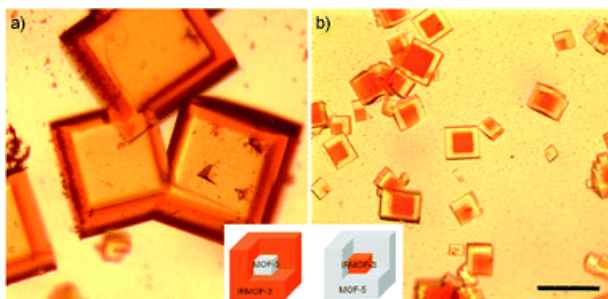
- Phosphorous-Nitrogen core lends excellent stability and good interaction with  $\text{CO}_2$ .
- Molecular foliage used to control molar volume and add additional  $\text{CO}_2$  affinity.
- Initial results do not appear favorable.

Poster

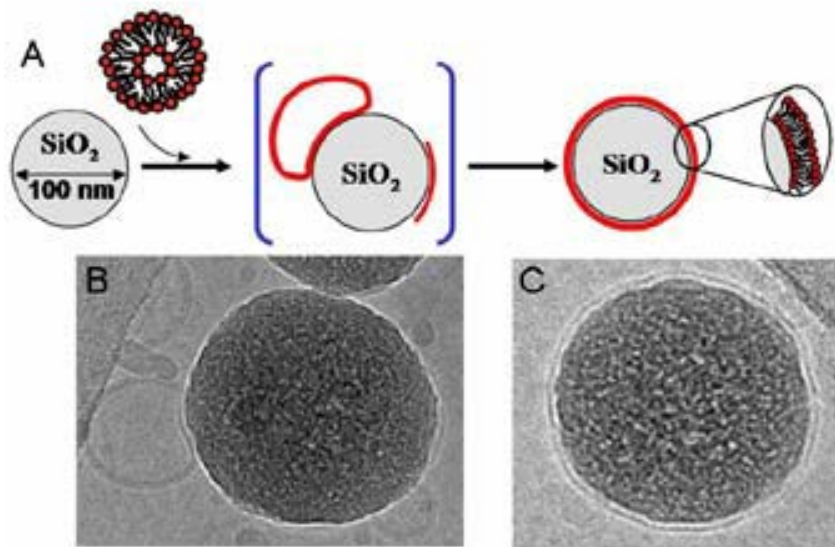
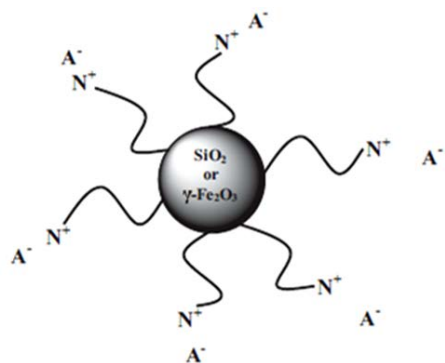
# Core-shell MOFs



- Many MOFs with desirable properties for CO<sub>2</sub> capture are water sensitive.
- It is possible to grow MOFs with similar crystal lattices in intimate contact with one another.
- MOF particles may be created with a core of CO<sub>2</sub>-philic, water sensitive MOF and a hydrophobic protective shell.
- Producing a practical capture material from the MOFs was not possible.



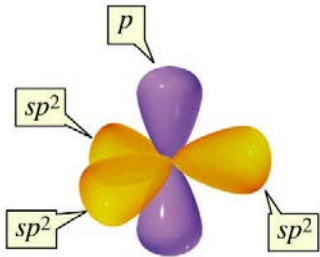
# Hybrid Nanoparticle Solvents



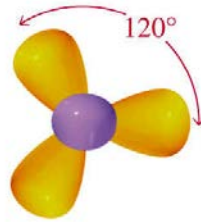
- Nanoparticles with appropriate ligands attached may behave as liquids.
- Depending upon the ligand and core, CO<sub>2</sub> capacity could be considerable.
- Cu-based nanoparticles of 10 nm size with targeted ligands were developed.
- Viscosity of the resulting liquids was too great, and they proved impractical.

Poster

# Porous Borazine-coated Silica

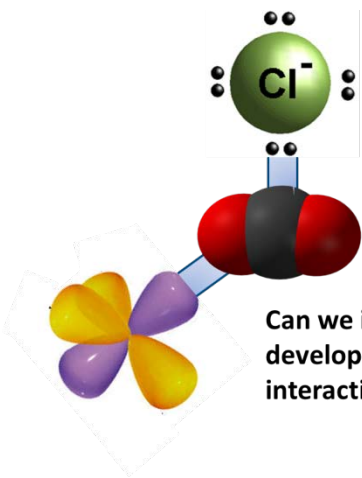


side view



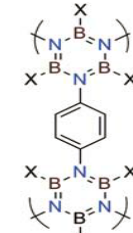
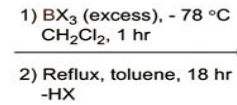
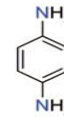
top view

- Boron containing materials have high  $\text{CO}_2$  uptake and good stability.
- Capture materials may be made by supporting them on high surface area materials.
- Material fabrication underway.

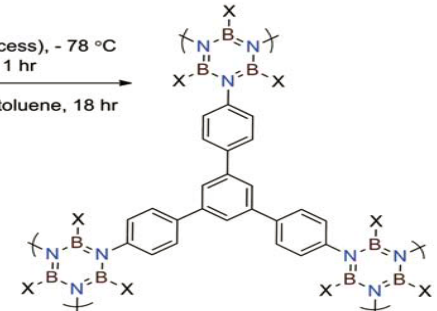
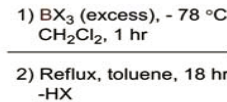
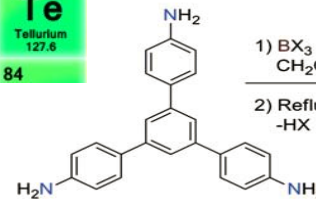


Can we improve  $\text{CO}_2$  interaction by developing a Bidentate type interaction?

13 IIIA 3A	14 IVA 4A	15 VA 5A	16 VIA 6A
5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.00674	8 O Oxygen 15.9994
13 Al Aluminum 26.981539	14 Si Silicon 28.0855	15 P Phosphorus 30.973762	16 S Sulfur 32.065
31 Ga Gallium 69.732	32 Ge Germanium 72.64	33 As Arsenic 74.92159	34 Se Selenium 78.96
49 In Indium 114.818	50 Sn Tin 118.71	51 Sb Antimony 121.760	52 Te Tellurium 127.6
81	82	83	84



BLP-1(Cl), 94%  
 BLP-1(Br), 78%



BLP-2(Cl), 80%  
 BLP-2(Br), 71%

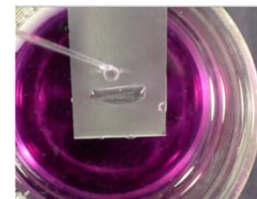
# Flexible Inorganic Polymer Membranes

## Polyphosphazenes:

High performance elastomers (aerospace)



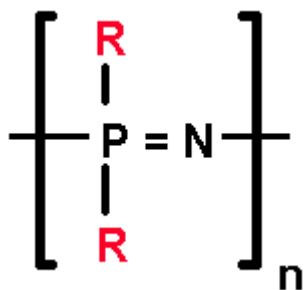
Surface modification by plasma or chemical etching



Hydrophobic fibers

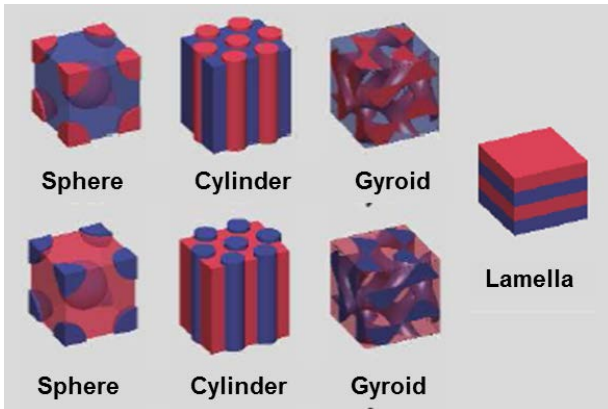


Contact angle of water droplets, over 100°



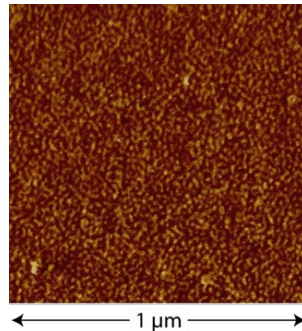
- Phosphorous-nitrogen backbone shows good CO<sub>2</sub> affinity.
- Several new polymers synthesized with properties targeted for CO<sub>2</sub> separations.
- Film formation in progress.

# Structured Polymers (PIs)



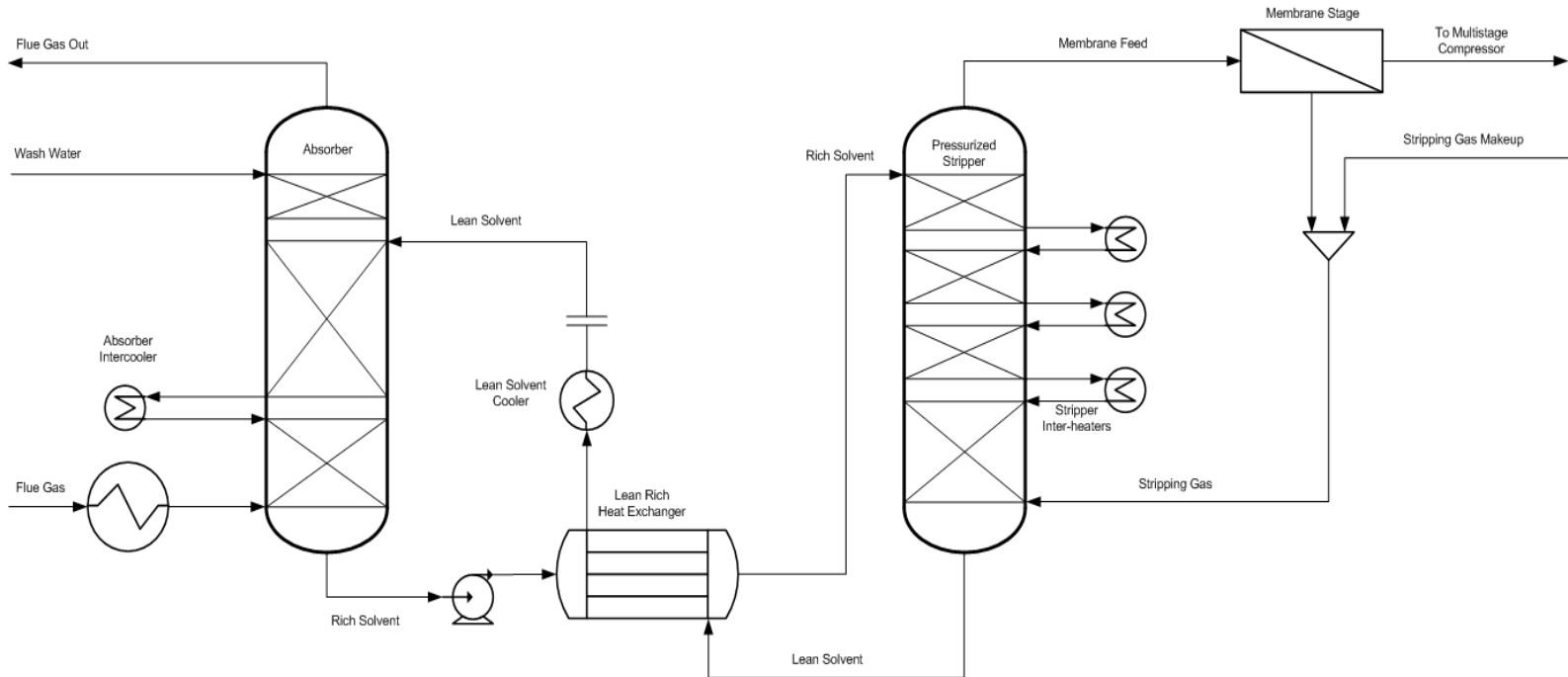
- Plasticization is a problem in CO<sub>2</sub>-selective membranes.
- Block copolymers can phase segregate at the nanoscale to produce separate domains.
- The property can be used in a membrane with separate transport and structural phases.
- Membrane films have been created using poly(IL)s as the transport phase.

**Poster**



# Processes

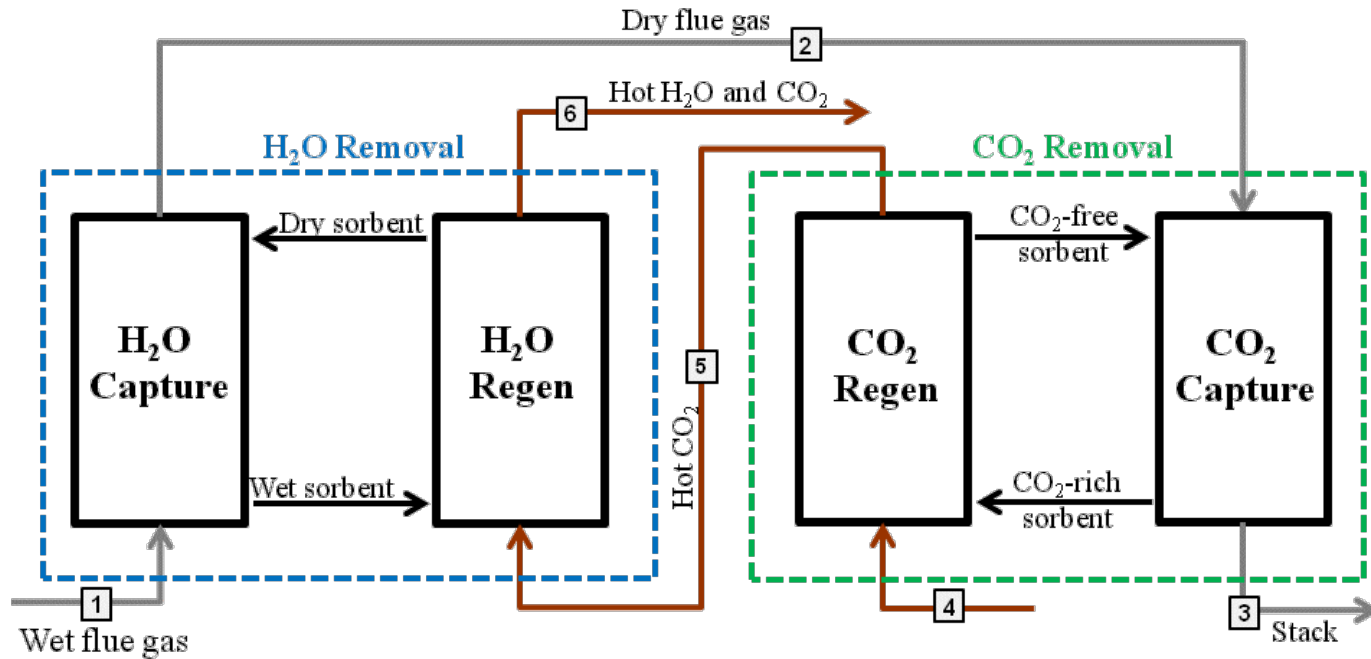
# Solvent-Membrane Hybrid System



- An amine solvent cycle may be used with a sweep gas to compress the CO<sub>2</sub> in the flue gas to higher pressure and concentration.
- A membrane can then be used to produce pure CO<sub>2</sub> ready for sequestration.
- Systems analyses were performed examining the economics of the process and it was found to be competitive with existing processes without materials development.
- Membrane development was undertaken to improve the process.



# Integrated Water Removal



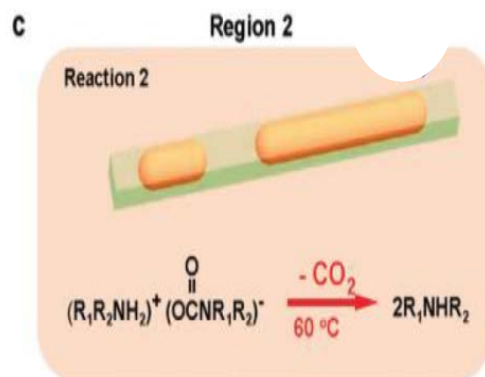
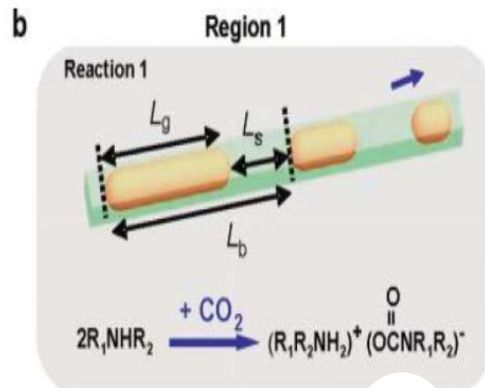
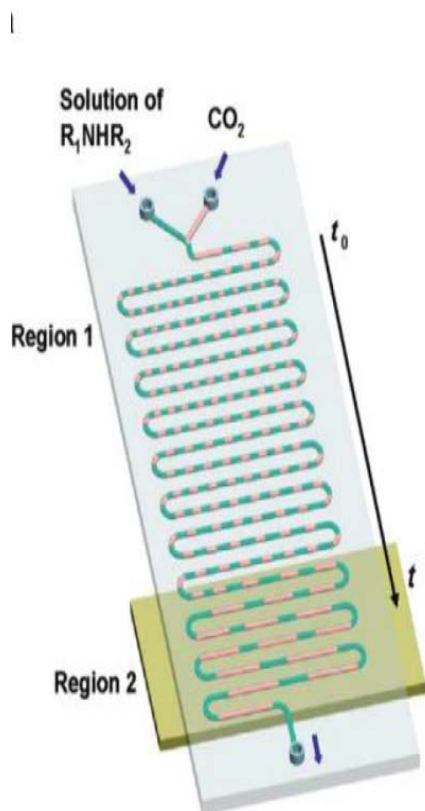
- Many potentially useful CO<sub>2</sub> capture techniques are infeasible because to the presence of water in flue gas.
- Using a low energy physical adsorption and making use of residual heat, the water may be removed concurrent to capture.

- Systems analyses were performed examining the economics of the process and it was found to be potentially competitive based on the capture technology used.

Poster

# Methods

# Microfluidic Apparatus for Solvent Characterization

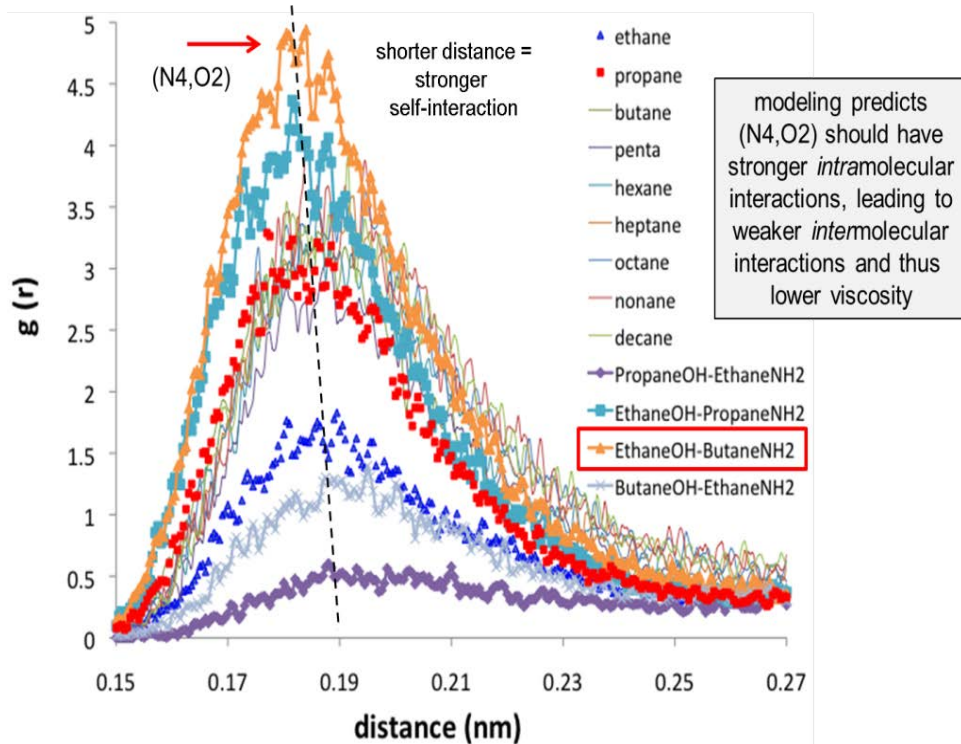
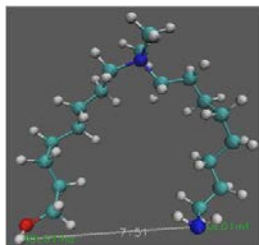
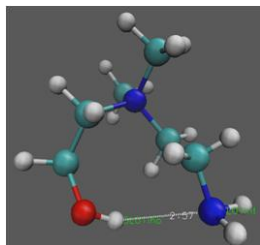


- Screening of solvents can require large volumes of material and substantial time commitment.
- A microfluidic device was developed which examines bubble shrinkage over time in contact with a liquid solvent to determine gas solubility and mass transfer rate.

# Conclusions

- **NETL-ORD uses an integrated technology development approach which examines a large number of technologies to determine their promise for CO<sub>2</sub> capture.**
- **A variety of materials, processes, and testing methods have been evaluated for their ability to achieve long term CO<sub>2</sub> capture targets.**
- **Some of the technologies show promise and further evaluation will be conducted.**

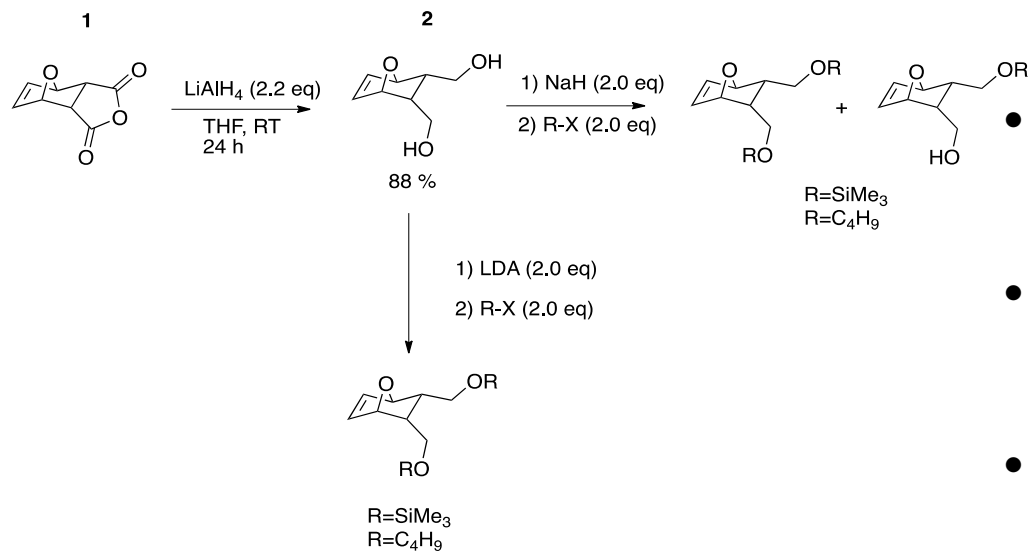
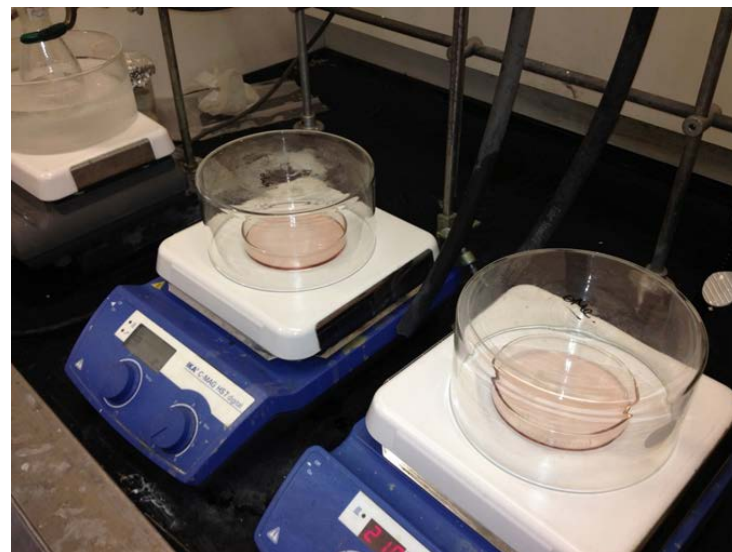
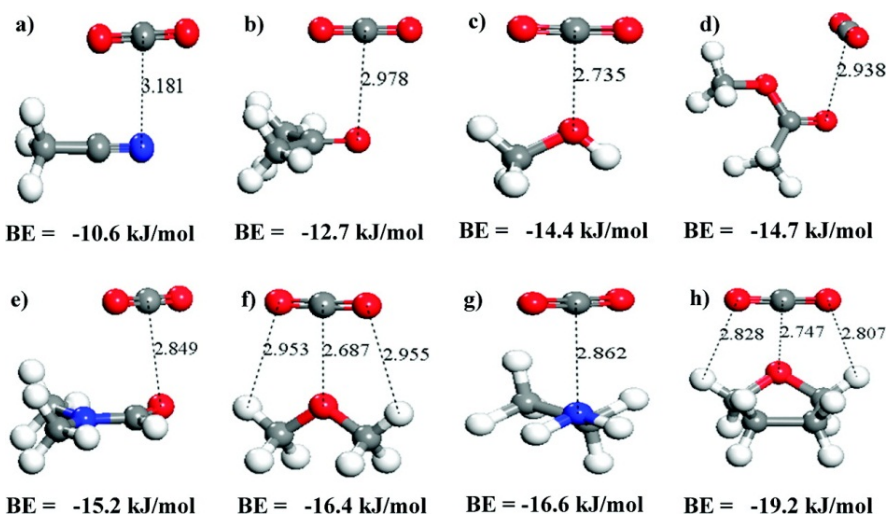
# Choline-based ILs



RFIL	viscosity cP	T <sub>g</sub> °C	Setaram CO <sub>2</sub> uptake, mol CO <sub>2</sub> / mol IL
[NH <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> NMe <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> OH]Tf <sub>2</sub> N (N2, O2)	4530	-39.4	0.017
[NH <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> NMe <sub>2</sub> (CH <sub>2</sub> ) <sub>3</sub> OH]Tf <sub>2</sub> N (N2, O3)	1146	-44.6	not determined
[NH <sub>2</sub> (CH <sub>2</sub> ) <sub>3</sub> NMe <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> OH]Tf <sub>2</sub> N (N3, O2)	1303	-49.6	not determined
[NH <sub>2</sub> (CH <sub>2</sub> ) <sub>3</sub> NMe <sub>2</sub> (CH <sub>2</sub> ) <sub>3</sub> OH]Tf <sub>2</sub> N (N3, O3)	1424	-39.2	0.018
[NH <sub>2</sub> (CH <sub>2</sub> ) <sub>3</sub> NMe <sub>2</sub> (CH <sub>2</sub> ) <sub>5</sub> CH <sub>3</sub> ]Tf <sub>2</sub> N (N3, hex)	1084	-46.8	0.018
[NH <sub>2</sub> (CH <sub>2</sub> ) <sub>4</sub> NMe <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> OH]Tf <sub>2</sub> N (N4, O2)	<b>280</b>	<b>-66.6</b>	<b>0.028</b>

- **Strong interactions with CO<sub>2</sub> are desirable for ILs as solvents and membranes.**
- **Inter-molecular hydrogen bonding leads to increased viscosity and reduced mass transport.**
- **Intra-molecular hydrogen bonding, which may be encouraged with spacer groups allows for reduced viscosity.**

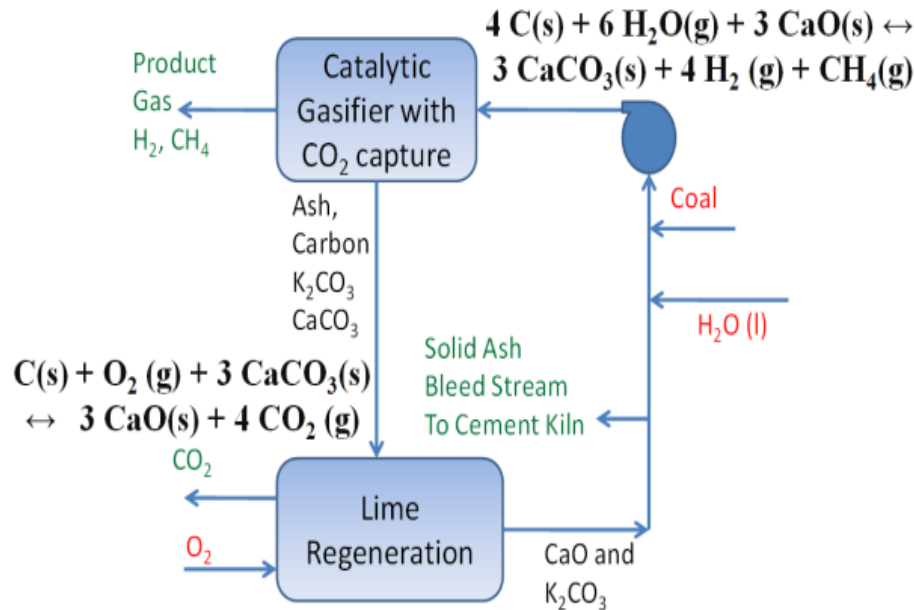
# Cyclic Ether-based Polymer Membranes



- Cyclic ether-based polymers have tailorable free volume and molecular affinity for CO<sub>2</sub>.
- Methodologies invented to synthesize monomers with desirable groups and polymerized these monomers.
- Film fabrication techniques developed for new materials.

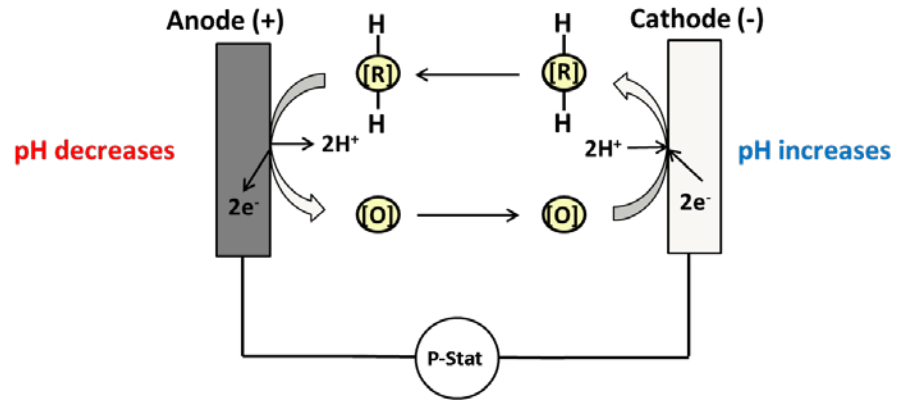
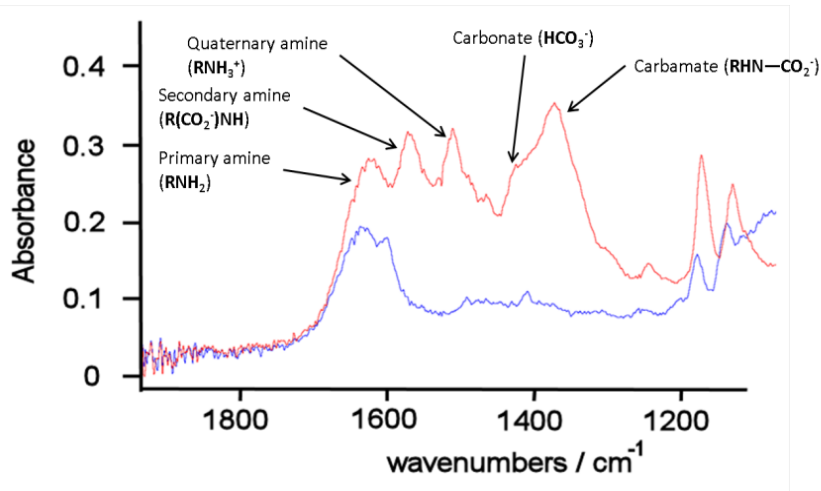
# Coal Gasification with In-Situ CO<sub>2</sub> Capture

Process Flow Diagram: Gasifier & Lime Regeneration



- Process uses alkali hydroxides and alkali earth metal oxides inside the gasifier as combined gasification catalysts and capture agents.
- The exothermic heat of reaction of the CO<sub>2</sub> capture reaction is utilized to offset the endothermic steam-coal gasification reactions.
- Lab scale coal gasification experiments and systems analyses are underway to examine the feasibility of the process.

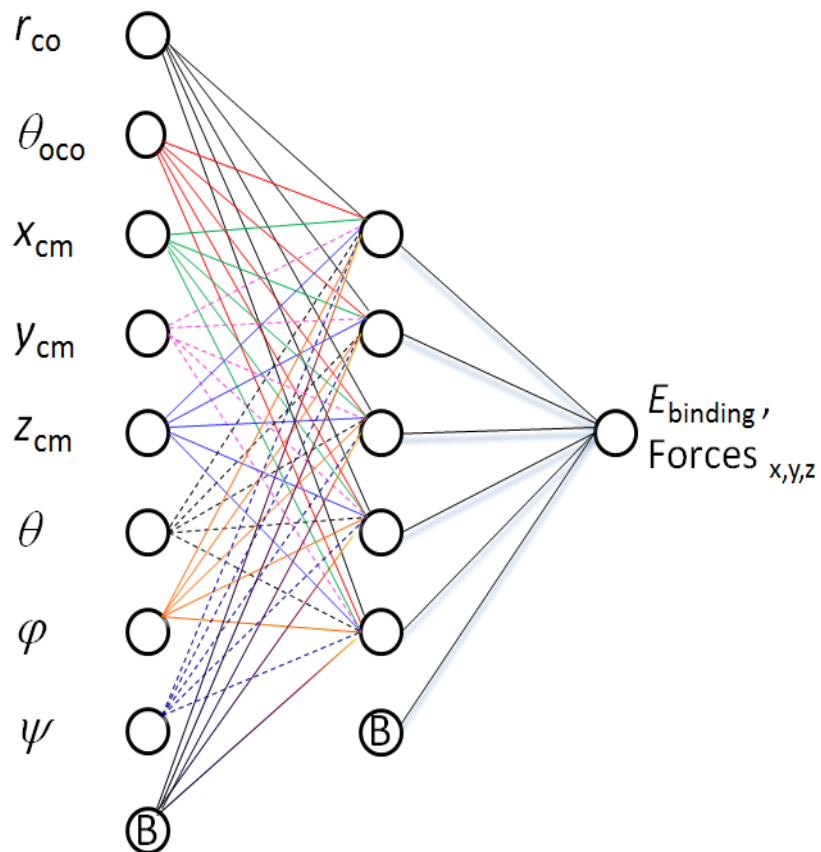
# Redox-driven Regeneration of Amines



- Electrically driven pH swing may be used to drive  $\text{CO}_2$  capture and solvent regeneration in a cycle based on quinone.
- A capture device has been constructed and results show  $\text{CO}_2$  concentration swing.
- Device may be employed in a membrane configuration.



# Neural Network Modeling



- NN is a processing system composed of a large number of highly interconnected processing elements.
- They work in unison to transform input data into output.
- Each neuron is defined by an activation function, which takes a weighted output of multiple input neurons as an argument.
- This input is used to train the force fields.
- An *ab initio* database for the specific case of  $CO_2$ - $[CH_3COO]^-$  anions is being generated.